

Enrollment No. _____



NAVVRACHANA UNIVERSITY
a UGC recognized University

School: School of Science
Program: M. Sc. in Chemistry (Analytical)
Year: 2nd **Semester:** 3rd
Examination: End Semester Examination
Examination year: December - 2021

Course Code: CH211 **Course Name:** SPECTROCHEMICAL ANALYSIS-I
Date: 02/12/2022 **Total Marks:** 40
Time: 2:30 to 4:30 pm **Total Pages:** 2

Instructions:

- Write each answer on a new page.
- Use of a calculator is permitted.
- * COs=Course Outcome mapping. # BTL=Bloom's Taxonomy Level mapping

Q. No.	Details	Marks	COs*	BTL#																																																							
Q.1	Match the following (Write complete option in the answer sheet)	10																																																									
	<table border="0"> <thead> <tr> <th>Sr. No.</th> <th>Column A</th> <th>Column B</th> <th></th> <th></th> </tr> </thead> <tbody> <tr> <td>1.</td> <td>calibration of IR instrument</td> <td>a) Scanning Tunneling Microscopy</td> <td></td> <td></td> </tr> <tr> <td>2.</td> <td>thermocouple</td> <td>b) selenium</td> <td></td> <td></td> </tr> <tr> <td>3.</td> <td>photovoltaic cell</td> <td>c) Attenuated Total Reflection</td> <td>CO1,</td> <td>BT1,</td> </tr> <tr> <td>4.</td> <td>diffraction grating</td> <td>d) SiC</td> <td>CO2,</td> <td>BT2</td> </tr> <tr> <td>5.</td> <td>ruby crystal</td> <td>e) epoxy resin</td> <td>CO4</td> <td></td> </tr> <tr> <td>6.</td> <td>globar</td> <td>f) ESR spectroscopy</td> <td></td> <td></td> </tr> <tr> <td>7.</td> <td>diamagnetic anisotropy</td> <td>g) polystyrene</td> <td></td> <td></td> </tr> <tr> <td>8.</td> <td>Curie point</td> <td>h) triglycine sulphate</td> <td></td> <td></td> </tr> <tr> <td>9.</td> <td>Constant height mode</td> <td>i) black body radiation</td> <td></td> <td></td> </tr> <tr> <td>10.</td> <td>evanescent radiation</td> <td>j) δ value</td> <td></td> <td></td> </tr> </tbody> </table>	Sr. No.	Column A	Column B			1.	calibration of IR instrument	a) Scanning Tunneling Microscopy			2.	thermocouple	b) selenium			3.	photovoltaic cell	c) Attenuated Total Reflection	CO1,	BT1,	4.	diffraction grating	d) SiC	CO2,	BT2	5.	ruby crystal	e) epoxy resin	CO4		6.	globar	f) ESR spectroscopy			7.	diamagnetic anisotropy	g) polystyrene			8.	Curie point	h) triglycine sulphate			9.	Constant height mode	i) black body radiation			10.	evanescent radiation	j) δ value					
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Q.2	Fill in the blanks	5																																																									
	<ol style="list-style-type: none"> The small constant response of a detector even in absence of radiation is called as ESR is also referred to as Wavelength corresponding to frequency 1720 cm^{-1} is The image formed by moving electron beam along X and Y axis is called as 		CO1, CO2, CO4	BT1, BT2, BT3, BT4, BT5																																																							

5. The region between 1500-650 cm^{-1} in IR spectrum is referred to as

Q.3 Answer the following

6

(a) Give full forms of ELDOR, NOE, COSY and DRIFTS.

CO2, BT1,
CO5 BT2,
BT5

(b) Give basic difference between NMR and ESR techniques.

(c) Explain mull and solid film techniques in IR spectroscopy.

Q.4 Explain any three of the following

9

(a) Atomic Force Microscopy

CO1, BT1,

(b) Job's Method of Continuous Variation.

CO2, BT2,

(c) Any three applications of ESR spectroscopy

CO4, BT3,

CO5 BT4

(d) Auger spectroscopy

Q.5 Answer any two of the following

10

(a) Sampling techniques for solid, liquid and gaseous samples in IR spectroscopy.

CO1, BT1,

CO2, BT2,

(b) Differentiate between Raman and infrared spectroscopy.

CO3, BT3,

CO5 BT4,

(c) DEPT technique for structure elucidation using NMR.

BT5

*****End of Question Paper*****