

Infrared spectroscopy of germanium dioxide (GeO₂) glass at high pressure

P. V. Teredesai,^{1*} D. T. Anderson,¹ N. Hauser,¹ K. Lantzky²
& J. L. Yarger^{3*}

¹Department of Chemistry, University of Wyoming, Laramie, WY 82071, USA

²Department of Chemistry, St. John Fisher College, Rochester, NY 14618, USA

³Department of Chemistry and Biochemistry, Arizona State University, Tempe, AZ 85287-1604, USA

Room temperature infrared studies have been performed on GeO₂ glass under quasi-hydrostatic pressures of 9.5 GPa. The mid-infrared absorption spectrum of GeO₂ glass (g-GeO₂) shows two distinct vibrational modes, one at 560 cm⁻¹ (mid frequency peak) and another at 870 cm⁻¹ (high frequency peak). At pressures above 6 GPa, these two absorption peaks broaden and intensity in the region between them (~700 cm⁻¹) increases dramatically. We interpret this change in the infrared spectrum at high pressure as the onset and progression of the amorphous to amorphous transformation from a low-density tetrahedral glass to a high density octahedral network glass.

Recently, many tetrahedral oxide glasses have been shown undergo a polyamorphic⁽¹⁾ or amorphous to amorphous phase transitions (AAT) at high pressure.⁽²⁾ Examples include glasses in the silicate, germanate and chalcogenide families as well as amorphous H₂O.⁽³⁾ High pressure studies of SiO₂ and GeO₂ glasses show evidence for an AAT above 30 GPa and 6 GPa, respectively,⁽³⁾ the new amorphous phase in both SiO₂ and GeO₂ being a high density six-coordinated glass.^(4–10)

While the glasses of amorphous silicates and germanates can undergo transitions to a new high density amorphous phase, so can the tetrahedral crystalline phase, which often shows a crystal-to-amorphous transition in a similar pressure region as the AAT seen in the amorphous materials.⁽¹¹⁾ Hence, either pressurising the ambient glass or tetrahedral crystalline material can be used to reach the high density amorphous phase in SiO₂ and GeO₂.

Amorphous GeO₂ is known to respond to pressure either through modifications of the medium range order⁽¹²⁾ or through coordination changes of the cations in the tetrahedral network.^(7–10) High pressure Raman study⁽⁸⁾ carried on GeO₂ glass (g-GeO₂) indicated an

AAT between 6–12 GPa where the gradual coordination change was marked by a change in the pressure derivative of the frequency shifts of the main Raman band, broadening and weakening of this band, and the appearance of a low frequency feature at 240 cm⁻¹. A high pressure Raman study carried out on crystalline GeO₂ of quartz type (q-GeO₂),⁽⁹⁾ showed a crystal to amorphous transition with an onset at 7 GPa. These studies pointed to saturation in the pressure dependence of the frequency of the main A₁ (441 cm⁻¹) Ge–O–Ge symmetric stretching mode and a marked increase in the intensity of this mode near 7 GPa. Above 7 GPa, the intensity of A₁ mode dramatically decreases and a broad band on the high frequency side of this mode develops.⁽⁹⁾ Infrared study of q-GeO₂ carried to 30 GPa, shows a gradual amorphisation towards an almost 6 coordinated amorphous GeO₂ between 6–12 GPa.⁽¹³⁾

The average Ge–O distance, <Ge–O>, in both g-GeO₂ as well as q-GeO₂ was found to increase from 1.73 Å (tetrahedral structure) to 1.85 Å (octahedral structure) between 6–12 GPa.⁽⁷⁾ The wide range of pressure (6–12 GPa) where both coordinates exist indicated that the 4–6 coordination change was gradual. On decompression, only g-GeO₂ showed complete reversibility to four-fold coordination with a large hysteresis, having <Ge–O> around 1.73 Å at 2 GPa, whereas <Ge–O> in q-GeO₂ dropped down only to 1.8 Å, somewhere intermediate between the Ge–O distances of octahedral and tetrahedral structures. This observation forces one to think about two possibilities: on decompression from high pressures, (a) there is co-existence of 50% tetrahedral and 50% octahedral GeO₂ and (b) the high coordinated GeO₂ structure doesn't relax back to four-fold structure instead stabilises in the intermediate five-coordinated GeO₂ structure. Recent *in situ* x-ray and neutron diffraction measurements on g-GeO₂ have shown g-GeO₂ to undergo a metastable phase of average coordination number five between 6–12 GPa.⁽¹⁴⁾ As an analogy, the existence of five-coordinated Si cations has

* Corresponding authors. Email: pallavi.teredesai@gmail.com, Jeff.Yarger@asu.edu